



ESPECTROSCOPIA DE RESONANCIA MAGNÉTICA NUCLEAR

M. Victoria Gómez Almagro
17th July, 2019



INTRODUCTION

NMR INSTRUMENTATION

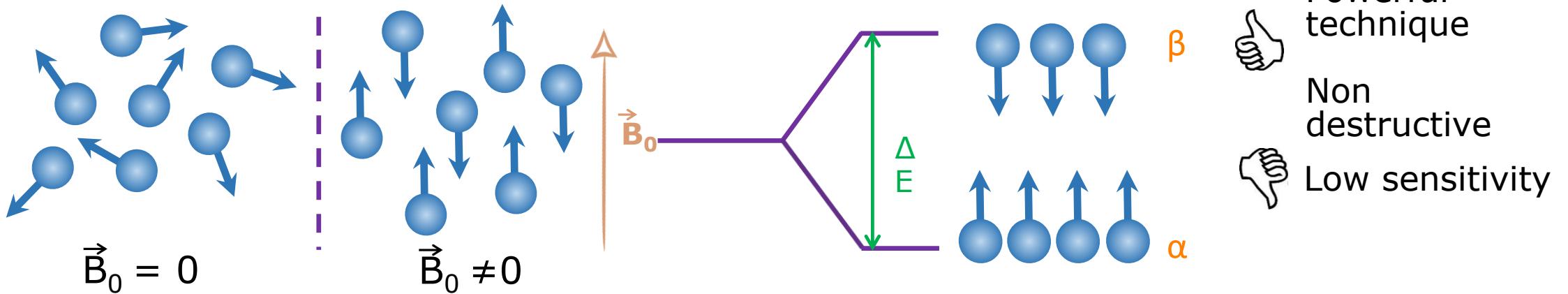
STEPS IN AN NMR ANALYSIS

STRUCTURAL IDENTIFICATION AND OTHER NMR APPLICATIONS

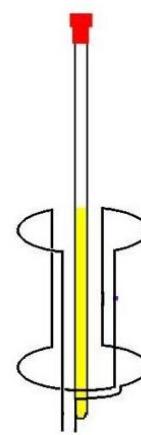
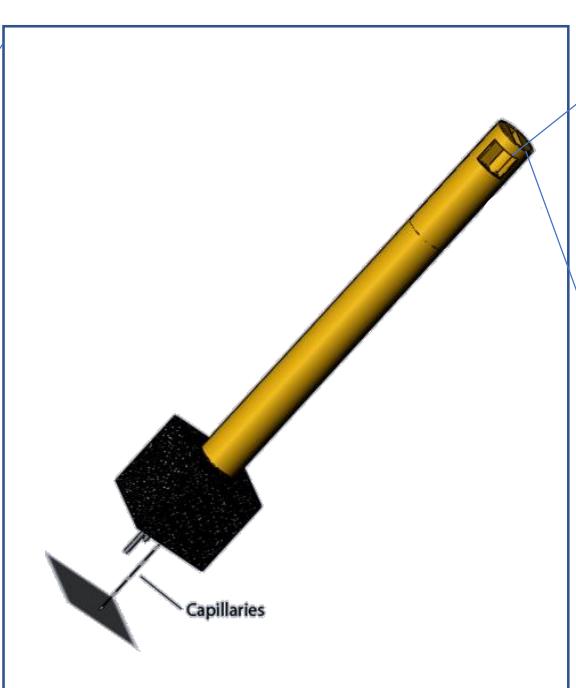
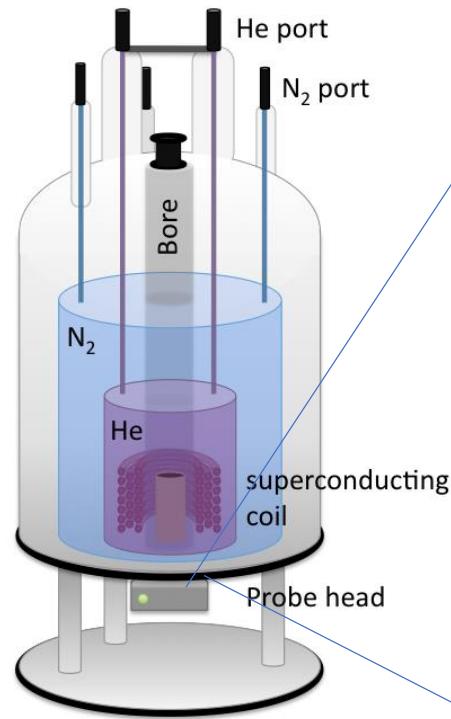
APPLICATIONS

- Structure identification.
- Reaction monitoring and kinetic investigations
- Reaction mechanism. Detection of intermediates
- Metabolomics
- Molecular size and aggregates: Diffusion NMR
- Monitorization of Dynamic process

INTRODUCTION



Powerful technique
Non destructive
Low sensitivity



OUR EQUIPMENT



500 MHZ



Triple resonance probe
TXI (500 MHz)



Higher ^1H sensitivity
More suitable for ^{15}N

www.bruker.com

iProbe
(400 y 500 MHz)



Universal.
Suitable for all nuclide

Different probes



Different applications

Every NMR analysis consists of:

NMR sample preparation

Liquid samples: 500 µL
>1 mg

(Probes for solid and semisolids samples)

Acquisition of NMR spectra



1D MULTINUCLEAR

2D MULTINUCLEAR

- $^1\text{H}, ^1\text{H}$ -COSY
- $^1\text{H}, ^1\text{H}$ -NOESY
- $^1\text{H}, ^{13}\text{C}$ -HSQC
- $^1\text{H}, ^{13}\text{C}$ -HMQC
- $^1\text{H}, ^{13}\text{C}$ -HMBC
- $^{19}\text{F}, ^{13}\text{C}$ -HSQC
- $^{19}\text{F} ^{13}\text{C}$ -HMBC

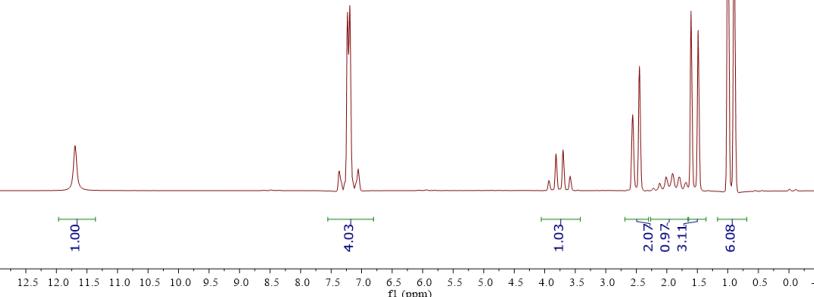
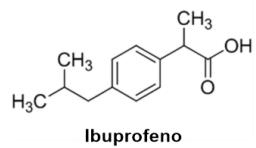
NMR assignment

Structure determination

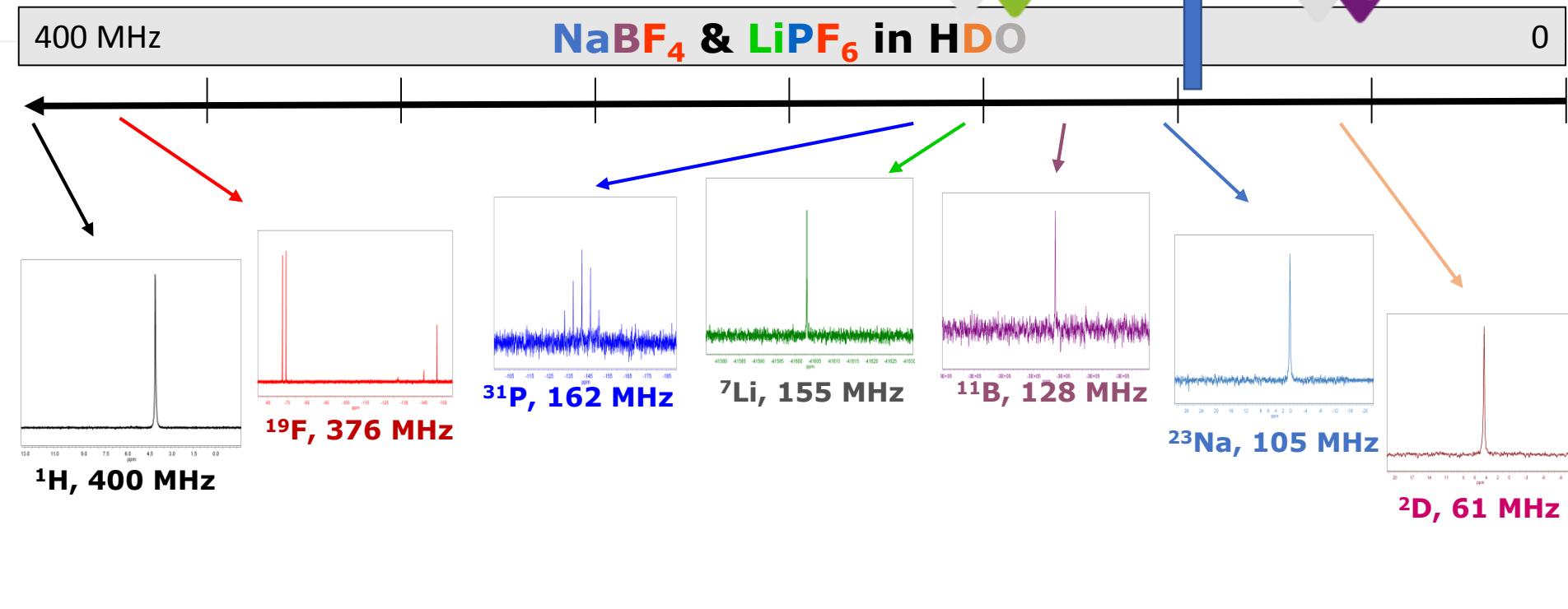
1D MULTINUCLEAR



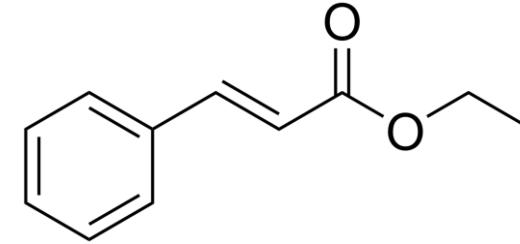
1D-1H-0-6.4-7-90



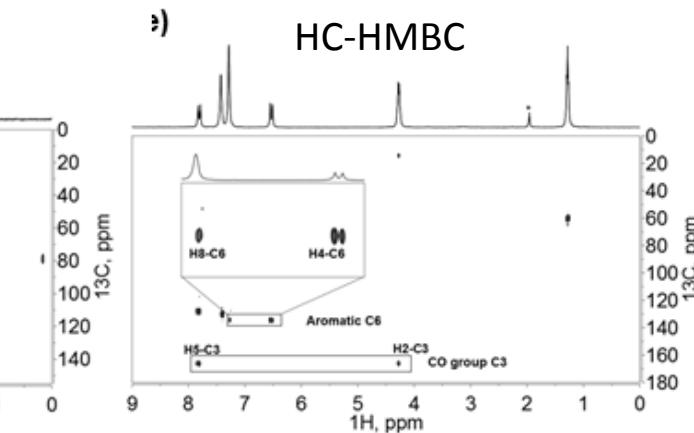
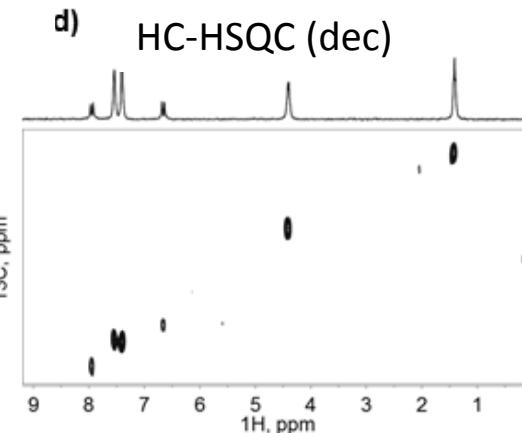
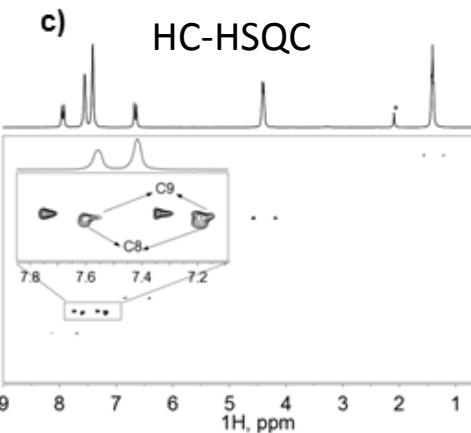
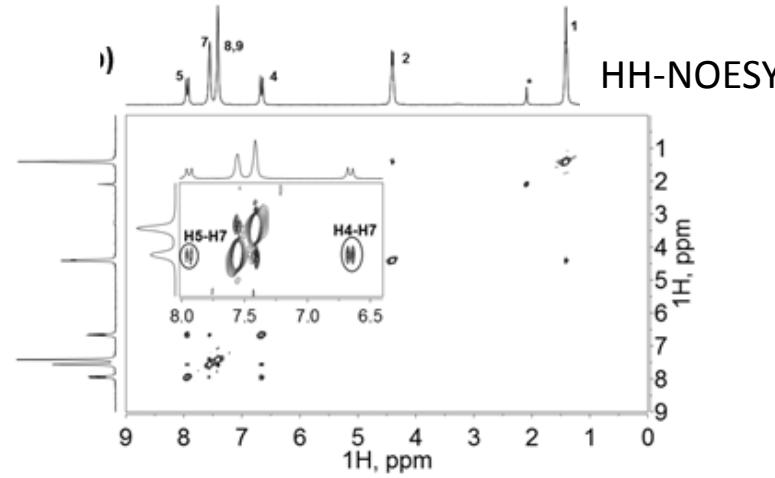
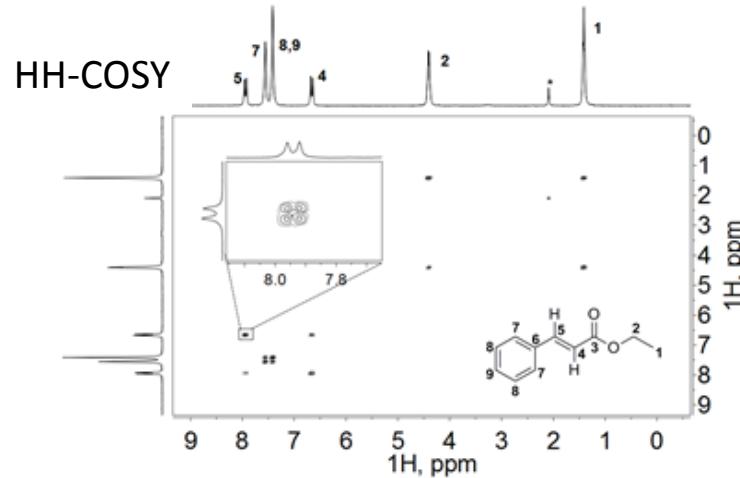
Looking at all frequencies



2D MULTINUCLEAR



ethylcinnamate (~ flavor in 1 pepernoot ...)



Every NMR analysis consists of:

NMR sample preparation

Liquid samples: 500 µL
>1 mg

(Probes for solid and semisolids samples)

Acquisition of NMR spectra



1D MULTINUCLEAR

2D MULTINUCLEAR

NMR assignment



PEAK IDENTIFICATION

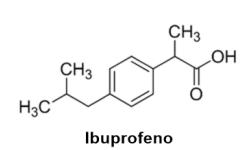
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Structure determination

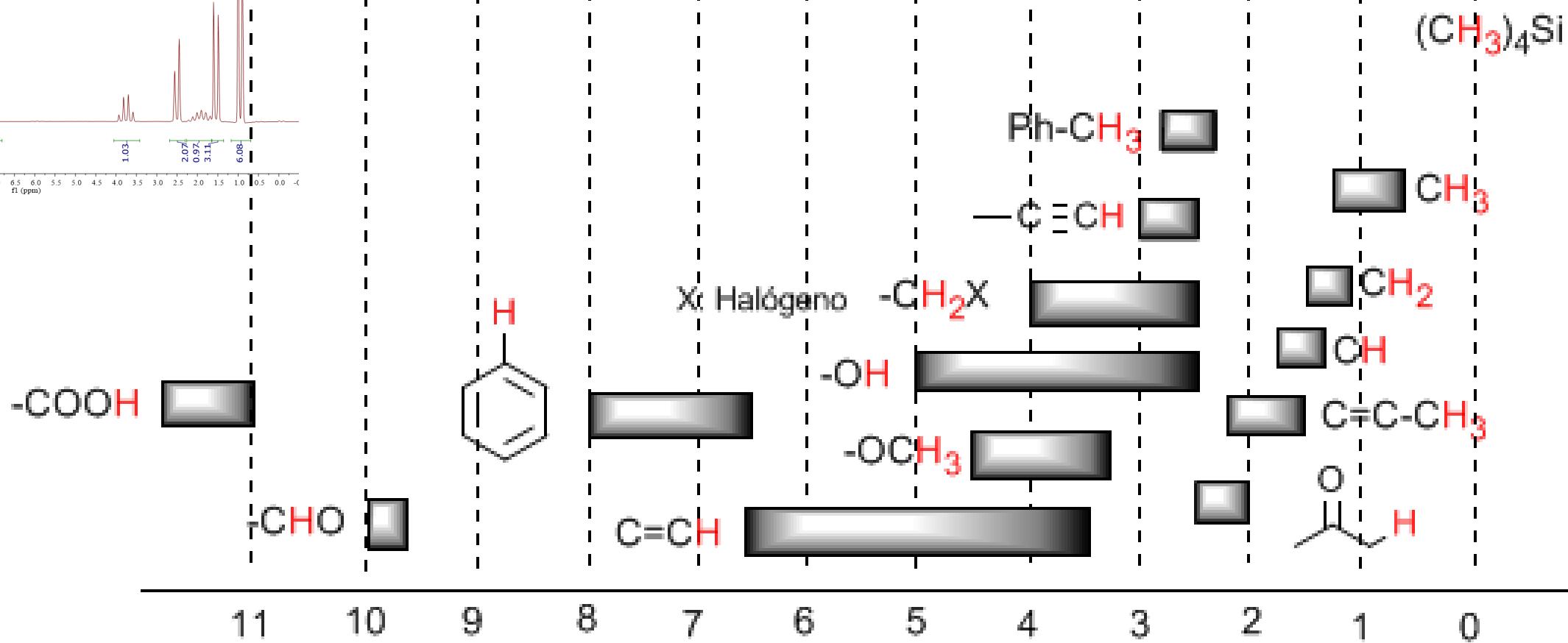
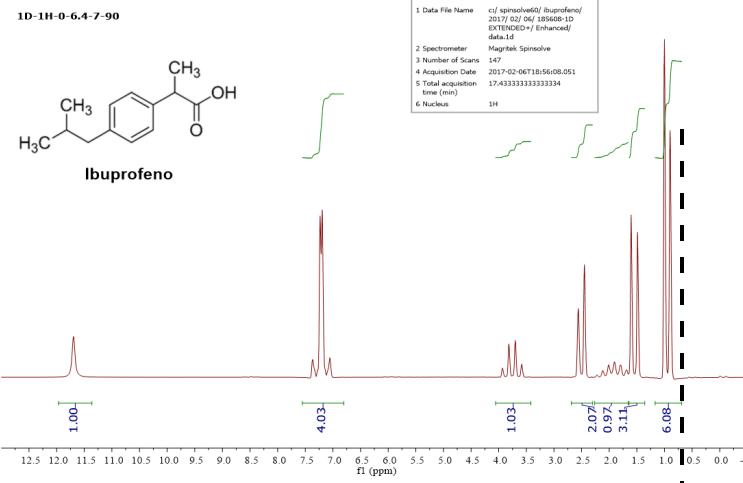
PEAK IDENTIFICATION

 magritek
1D-1H-9-6-4-Z-90

1D-1H-0-64-7-00



profeno



DESPLAZAMIENTOS QUÍMICOS EN ^1H RMN

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>1 mg

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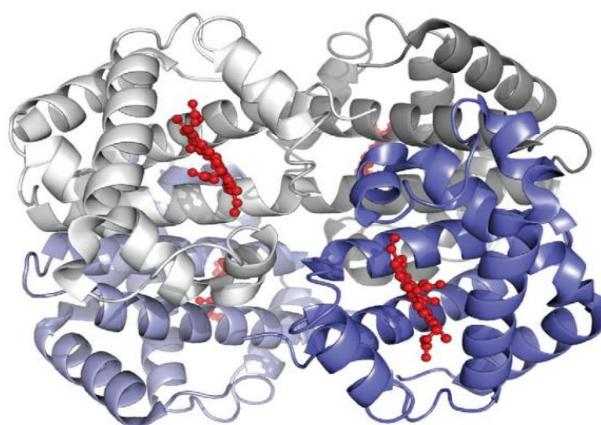
1D MULTINUCLEAR

NMR assignment



PEAK IDENTIFICATION

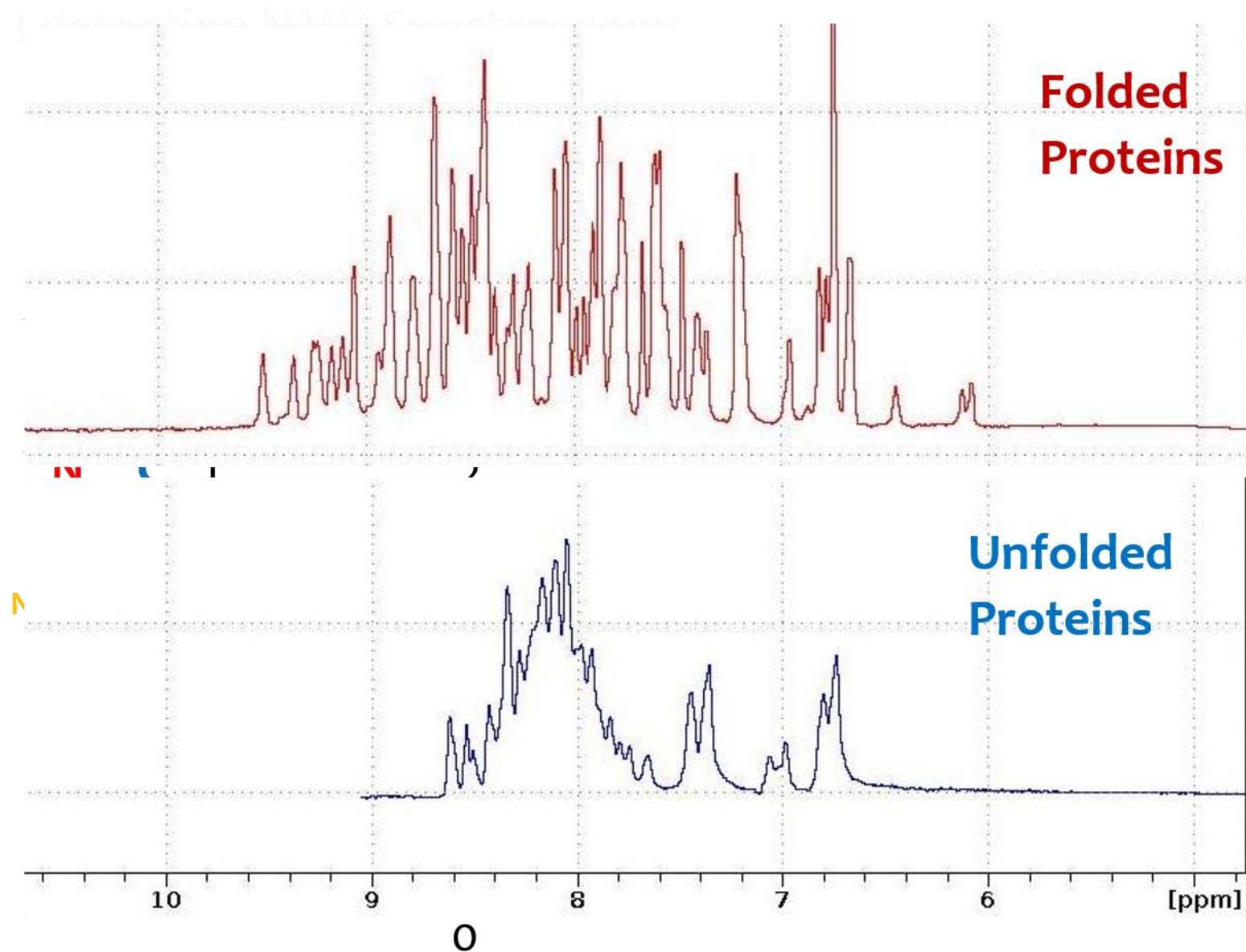
Structure determination



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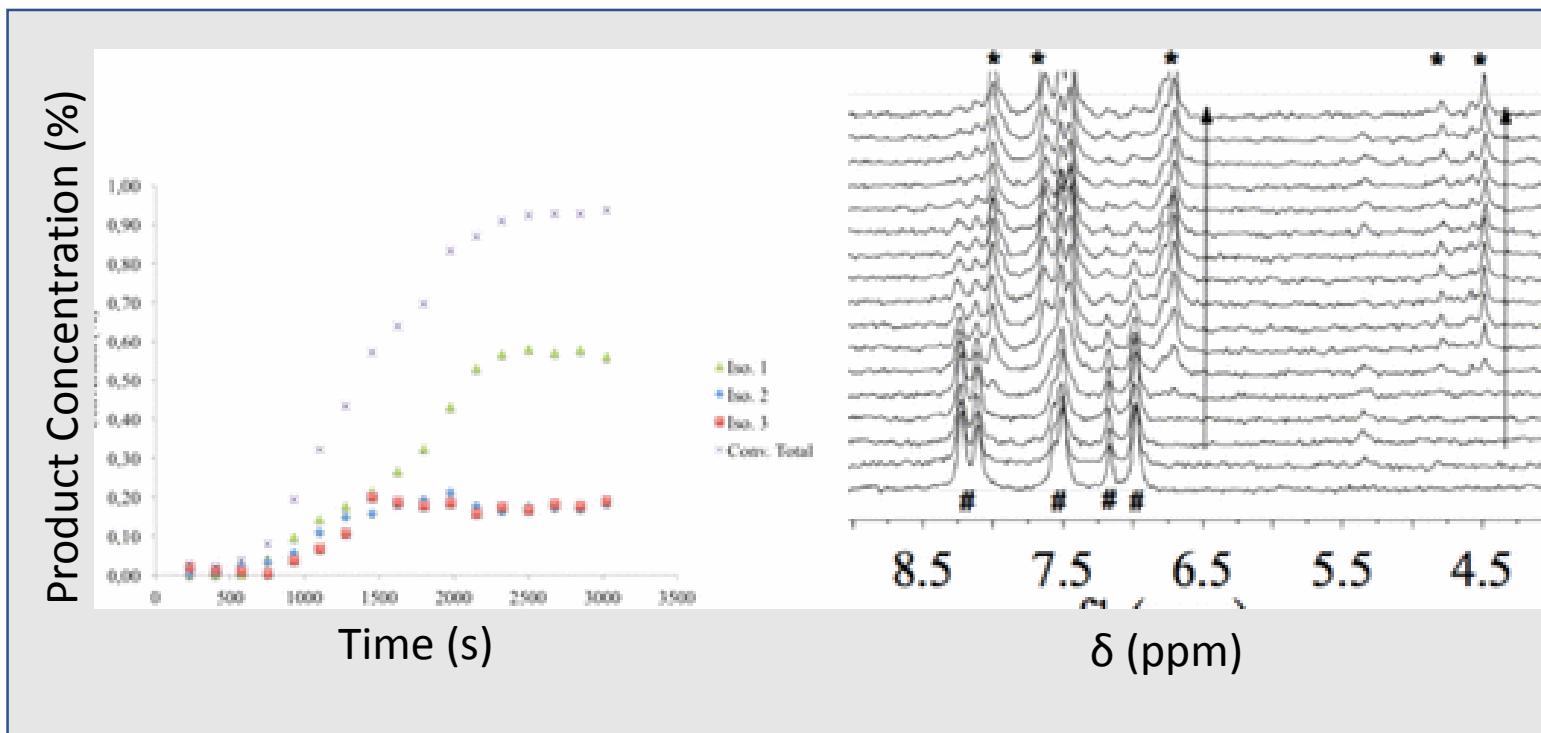
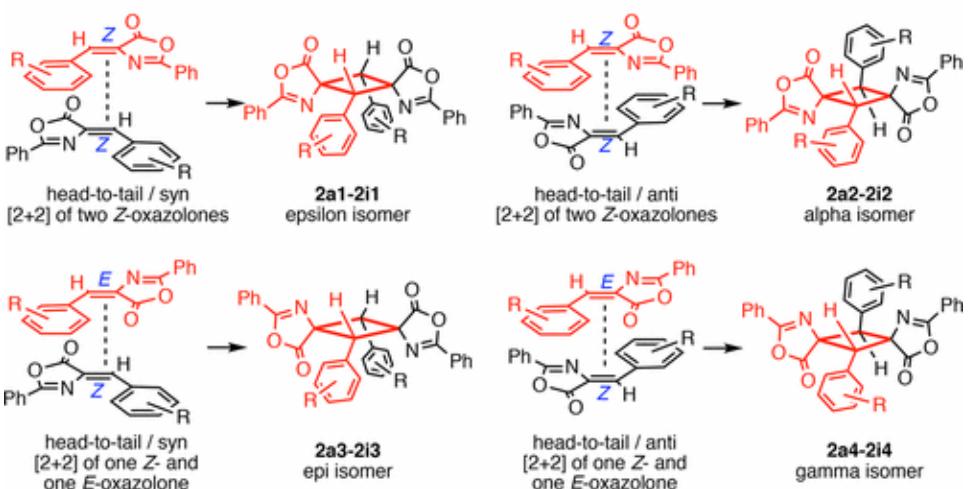
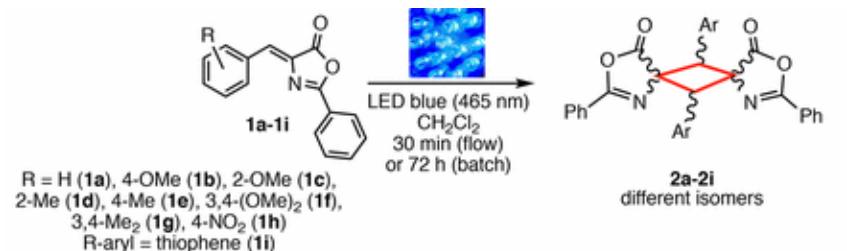
STRUCTURE IDENTIFICATION. i.e. PROTEINS



APPLICATIONS

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- Reaction mechanism. Detection of intermediates
- Metabolomics
- Molecular size and aggregates: Diffusion NMR
- Monitorization of Dynamic process

REACTION MONITORING. KINETIC INVESTIGATIONS



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MECHANISM ELUCIDATION. DETECTION OF INTERMEDIATES

Visible-Light-Induced Nickel-Catalyzed Negishi Cross-Couplings by Exogenous-Photosensitizer-Free Photocatalysis

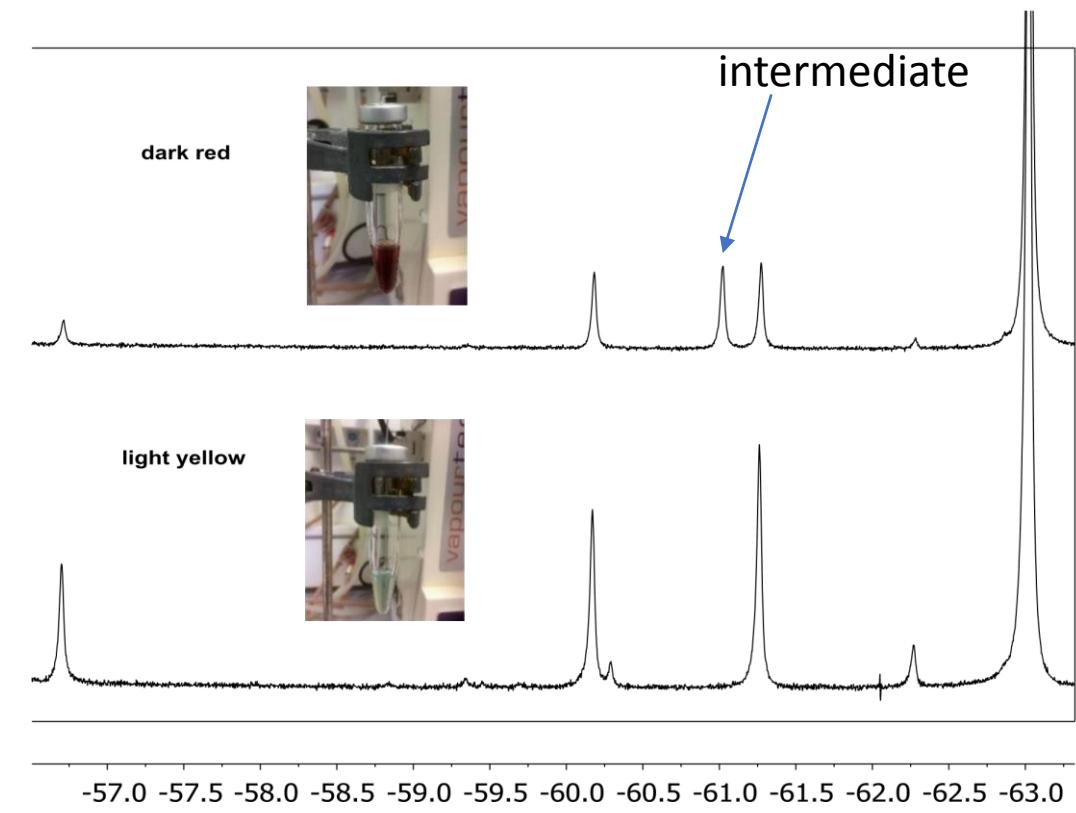
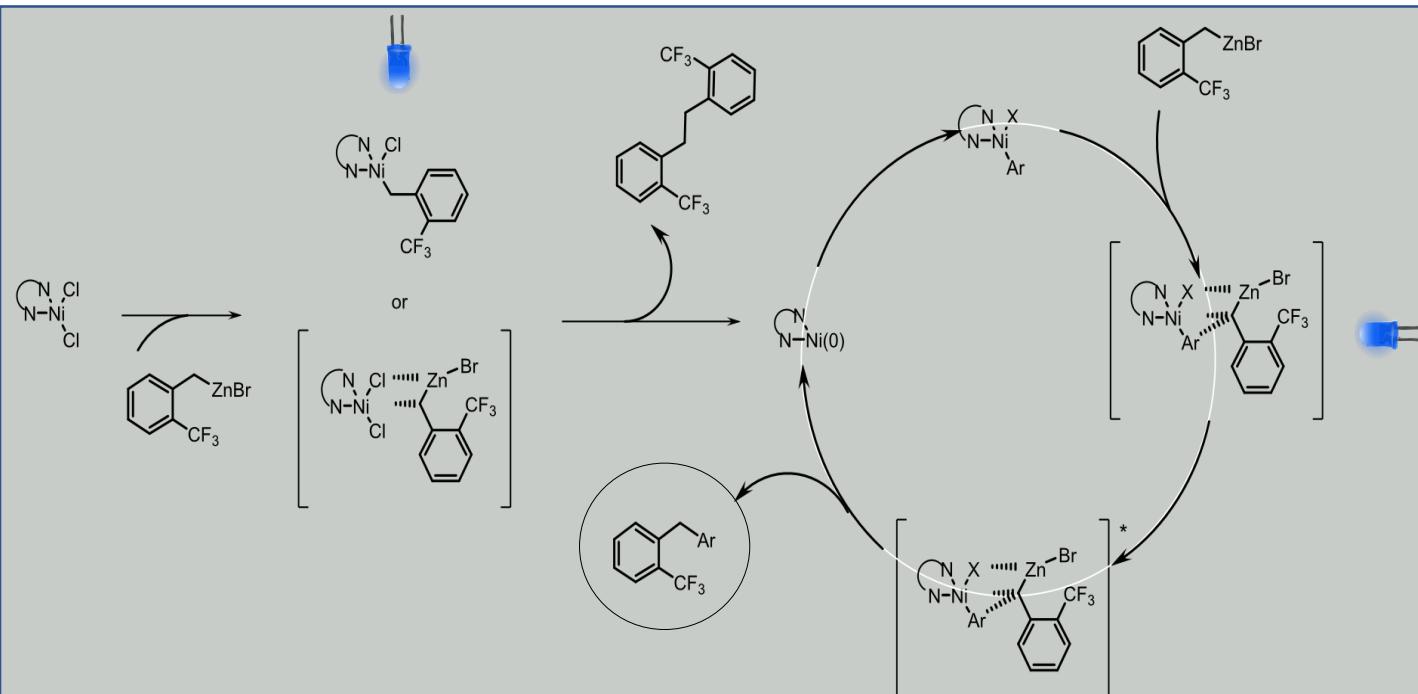


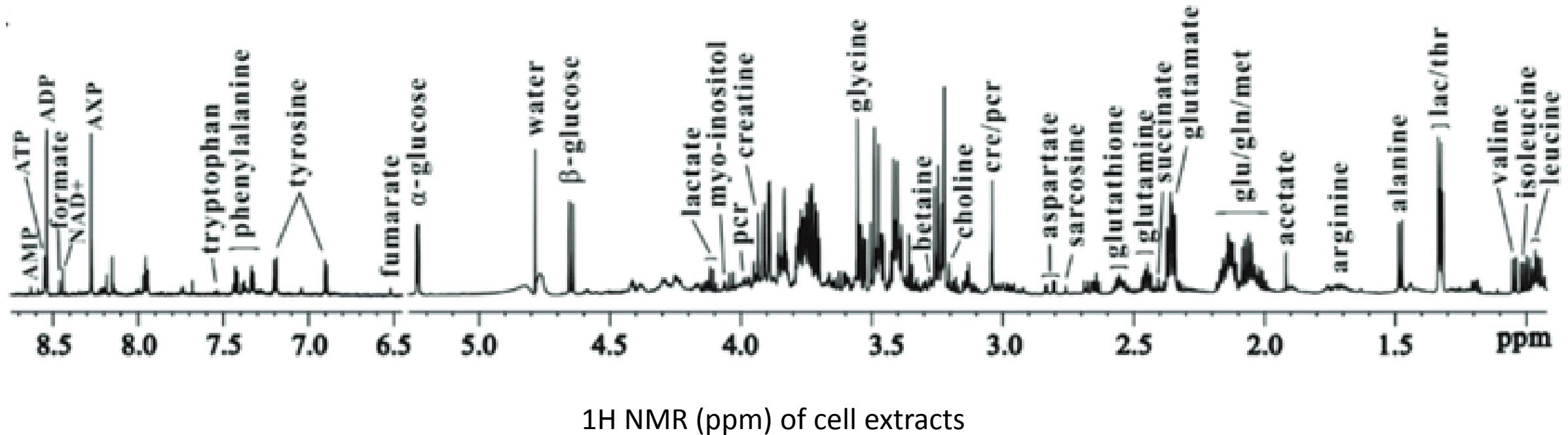
Figure S5: Top: ¹⁹F NMR spectrum after a quick addition of the zinicate derivative over the Nickel complex. The solution turned a dark red color. Bottom: ¹⁹F NMR spectrum of the reaction mixture within time and after light irradiation. The peak at -61 ppm has disappeared and the reaction media lost its dark red color.

APPLICATIONS

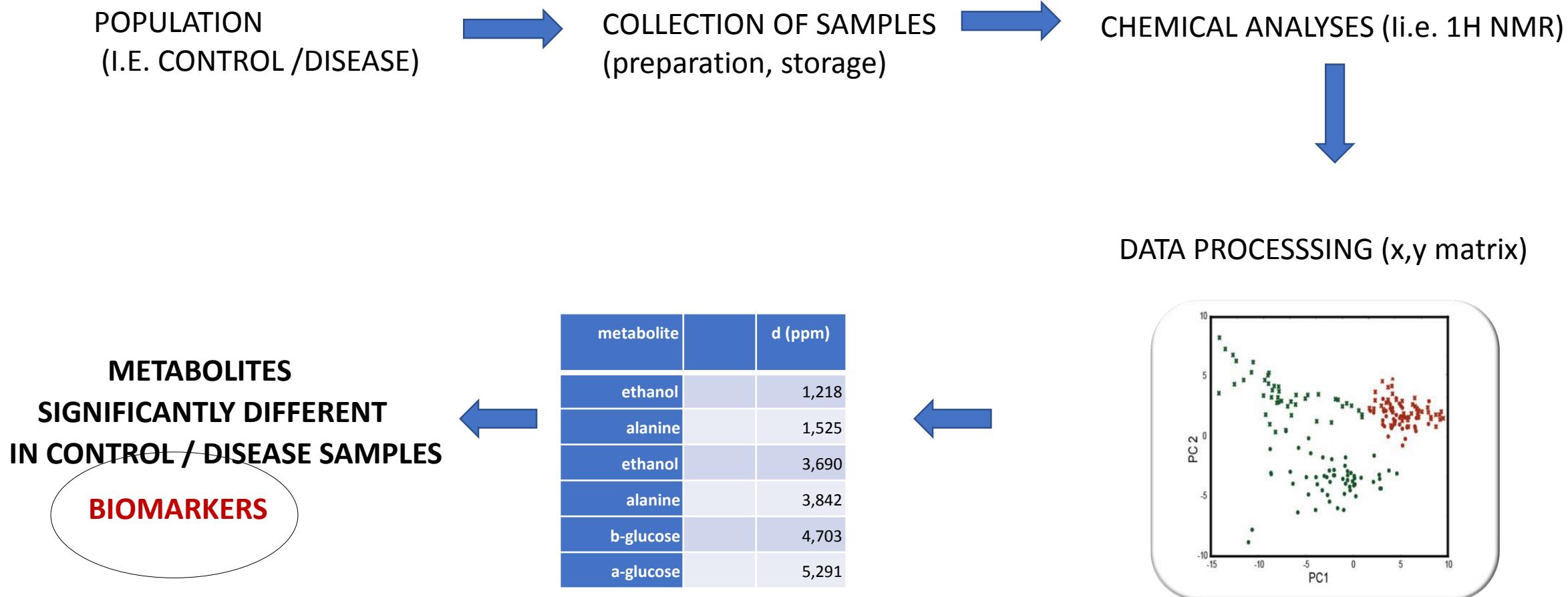
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METABOLOMICS

BLOOD PLASMA
URINE OR
CELL EXTRACTS



METABOLOMIC APPROACH



-43 metabolites identified in human Blood Plasma by 750 MHz ^1H and $^1\text{H}-^{13}\text{C}$ NMR spectroscopy:
Anal. Chem. **1995**, 67, 793-811.

APPLICATION OF METABOLOMICS

PHARMACEUTICAL DISCOVERY DISEASE



DIAGNOSIS AND DEVELOPMENT



ENVIRONMENTAL METABOLOMICS



PLANT METABOLOMICS



FOOD AND NUTRITION SCIENCE



ECOMETABOLOMICS



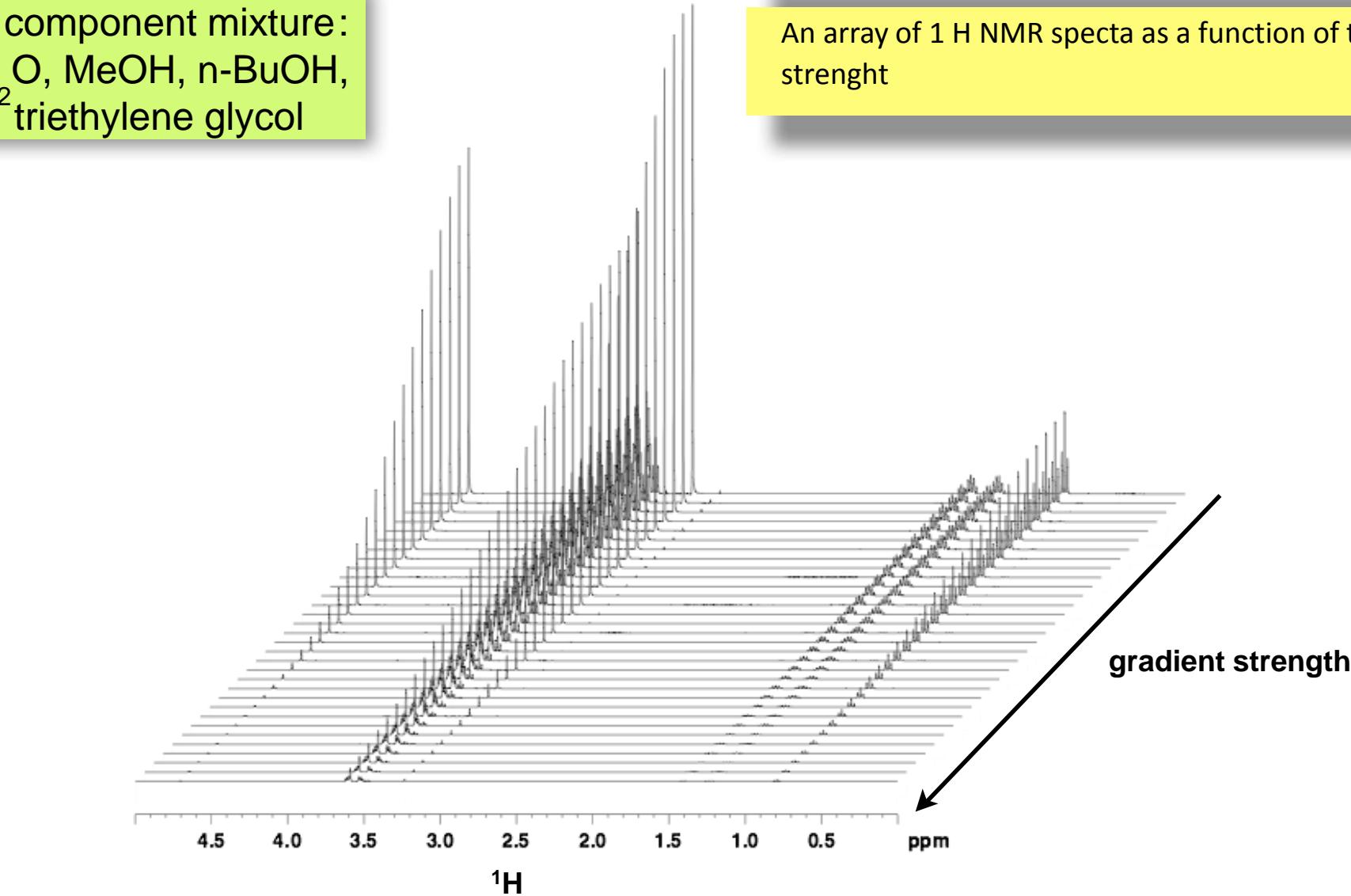
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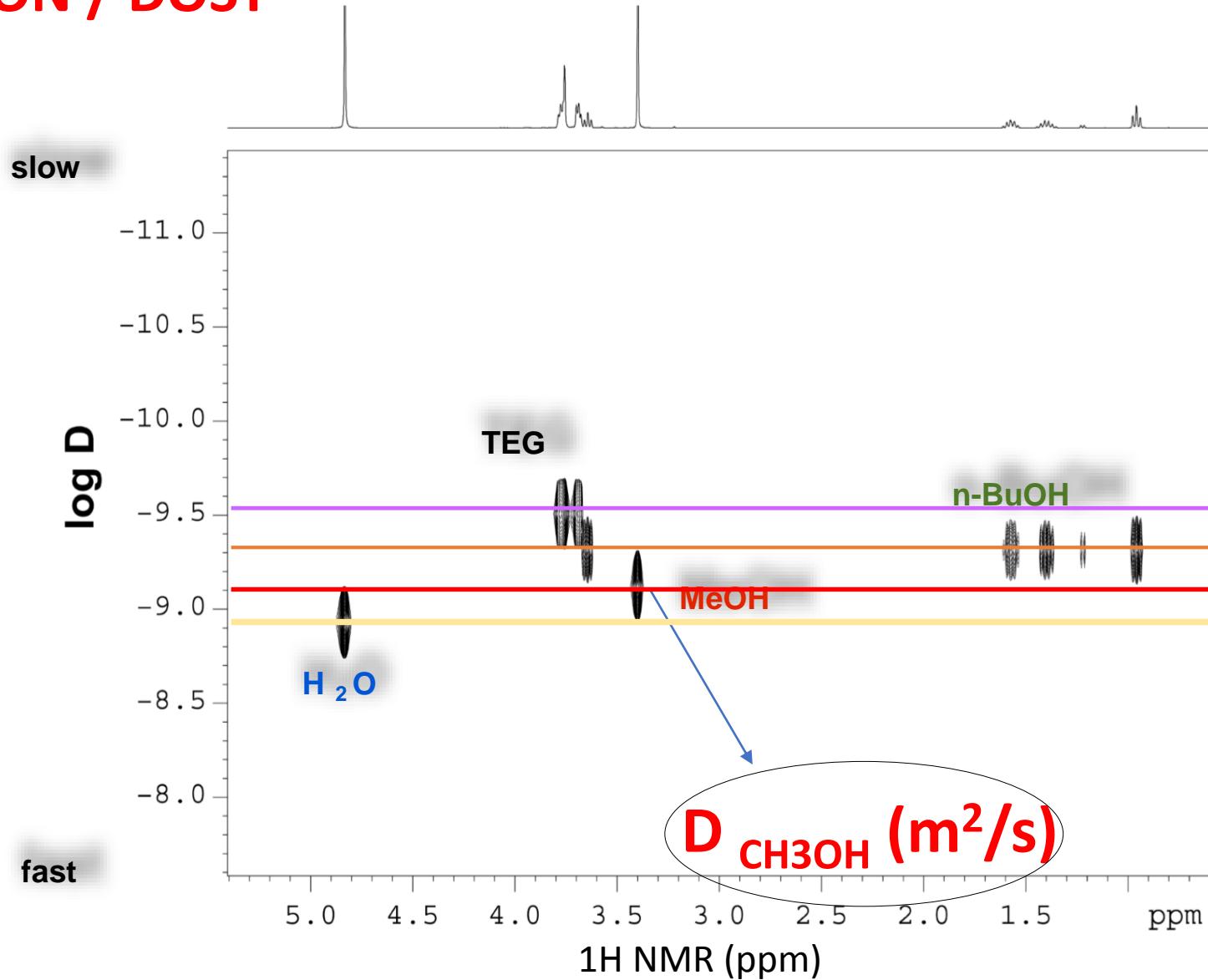
DIFFUSION / DOSY

4 component mixture:
 H_2O , MeOH, n-BuOH,
triethylene glycol

An array of ^1H NMR spectra as a function of the gradient strength



DIFFUSION / DOSY



Ratio in size:

TEG > n-BuOH > MeOH > H₂O

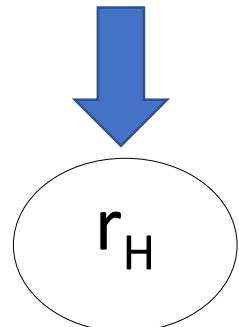
Quantitative diffusion studies

If the r_H of the standard is known the r_H of the sample can be determined

The ratio of the diffusion of a particular compound and the reference will be independent of the viscosity

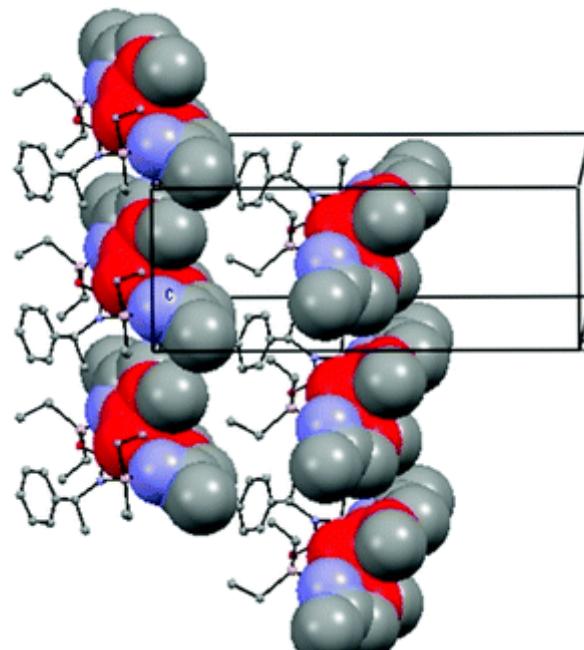
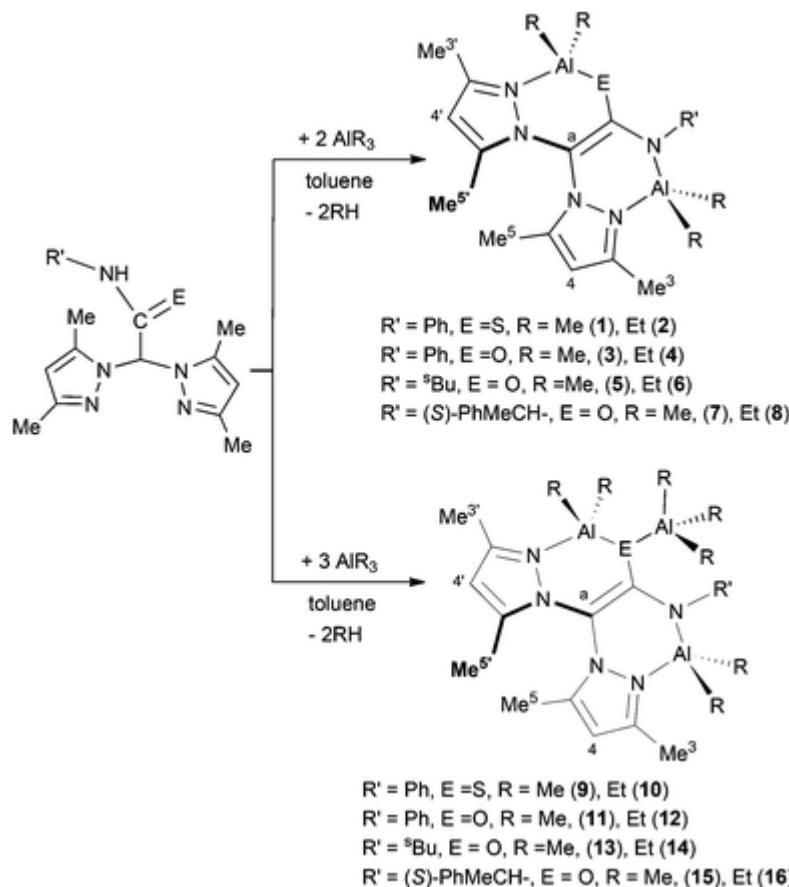
$$\frac{D}{D_{ref}} = \frac{r_{Href}}{r_H}$$

The use of a diffusion standard (D^{ref})



Heteroscorpionate aluminium complexes as chiral building blocks to engineer helical architectures

Dalton Trans., 2013, 42, 14240-14252

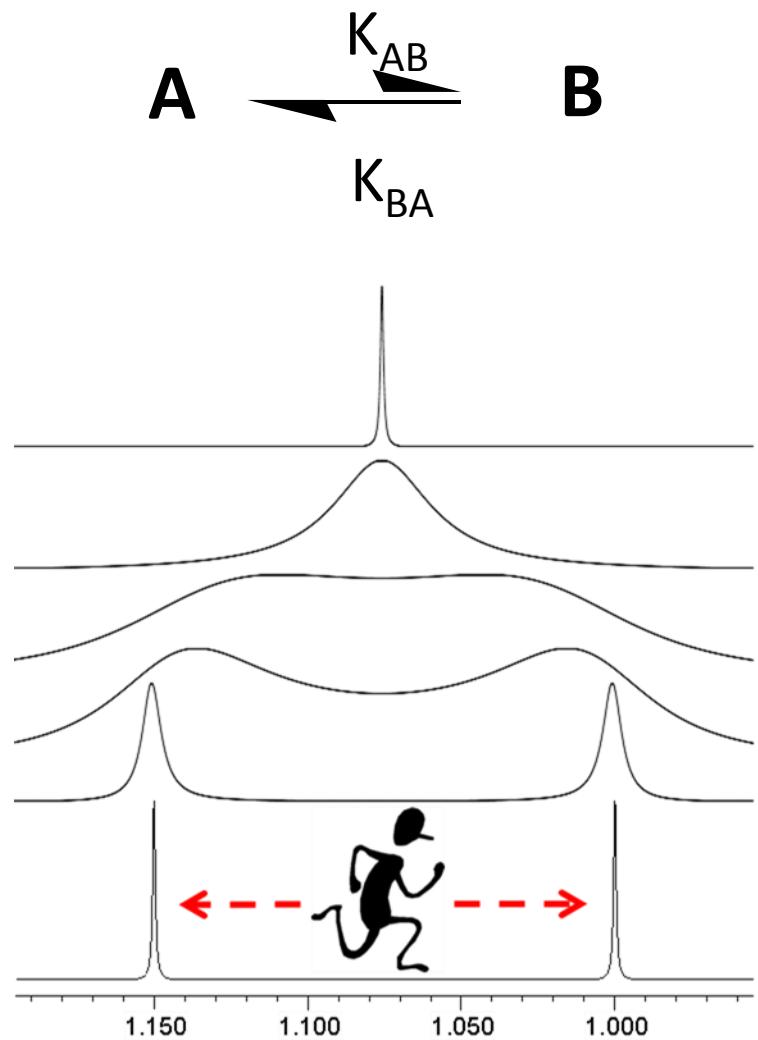


C (mM)	D _t (10 ⁻¹⁰ m ² /s)	r _H	V	N
10	10.4	5.46	0.6	1
50	8.3	6.85	1.3	1.90
100	7.05	7.46	1.7	2.55
300	5.9	8.0	2.1	3.20
500	4.8	8.6	2.7	3.93

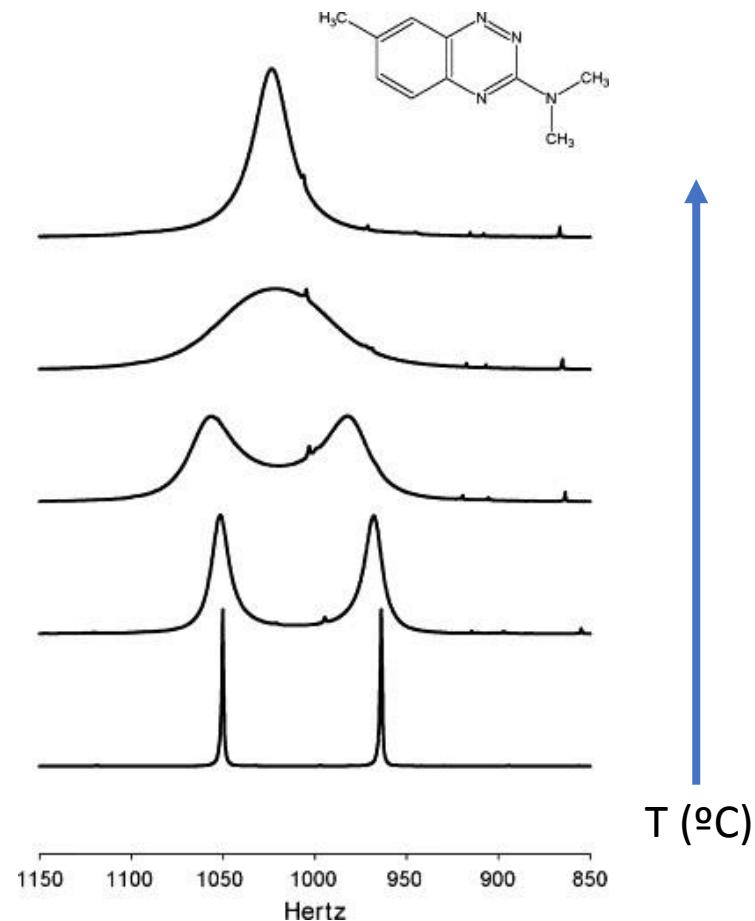
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DYNAMIC PROCESSES /CHEMICAL EXCHANGE



Ej. Typical example:
Variable temperature measurements



Other examples: axial-equatorial Exchange
keto-enol tautomerism



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